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## ARC AND SPARK SPECTRA OF YTTERBIUM

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## ABSTRACT

A new description of conventional arc and spark spectra of ytterbium (neoytterbium, aldeberanium) has been completed in the wave-length range 2000 to 11 000 Å. Wave-length measurements and intensity estimates are presented for 1,668 lines, 400 of which characterize neutral Yb atoms (Yb I spectrum), about 1,250 are due to singly ionized atoms (Yb II spectrum), and possibly a dozen belong to doubly ionized atoms (Yb III spectrum). Some spectral regularities are given for neutral and for singly ionized atoms. The ground states of Yb,  $\text{Yb}^+$ , and  $\text{Yb}^{++}$  atoms appear to be associated with the electron configurations,  $4f^{14}6s^2$ ,  $4f^{14}6s$ , and  $4f^{14}$ , respectively.

## CONTENTS

	Page
I. Introduction.....	651
II. Experiments.....	652
III. Results.....	653
IV. References.....	664

## I. INTRODUCTION

In 1878, Marignac [1]<sup>1</sup> discovered in gadolonite a new earth which he named ytterbium (Yb). Nearly 30 years later Urbain [2] succeeded in splitting ytterbium into two components, which he named neoytterbium (Ny) and lutecium (Lu). The same separation was made independently (and probably somewhat earlier) by Auer [3], who proposed the names aldeberanium (Ad) and cassiopeium (Cp). The common usage now is to represent the elements having atomic numbers 70 and 71 by the symbols Yb and Lu, respectively, except in Austria and Germany where Ad and Cp are more generally used.

Owing to the difficulty of separating rare earths from one another, and to the complex nature of their spectra, descriptions of the latter have long been in a most unsatisfactory state. In our recent paper [4] on the arc and spark spectra of lutecium, we remarked that all available samples of lutecium were contaminated by ytterbium and thulium so that the spectra of the former could not be unambiguously determined without simultaneous comparison with ytterbium and thulium spectra. Thus our attempt to produce a complete and trustworthy description of lutecium spectra naturally led to the production of similar data for thulium and ytterbium. Improved descriptions of these spectra are essential not only for purposes of spectroscopic identification and analysis, but also for the term analyses and interpretation or spectral structures characteristic of different

<sup>1</sup> Numbers in brackets refer to literature citations at the end of this paper.

atoms and ions. In the present paper we make public our latest description of the conventional arc and spark spectra of ytterbium, including some preliminary details about spectral structures.

The only spectroscopic tables published for ytterbium since the discovery of lutecium are the following [5]. Auer [3] first gave a list of 200 spark spectrum lines (2621.24 to 5897.56 Å), but most of the wave lengths were borrowed from a table which Exner and Haschek had given for old ytterbium containing unrecognized lutecium. In 1910, Eder and Valenta [6] published 58 lines (5474.24 to 5788.87 Å) observed in the arc spectrum and in 1911, reproduced spectrograms and gave tables of wave lengths [7] for the arc spectrum (169 lines 2615.48 to 6799.87 Å) and for the spark spectrum (195 lines 2615.48 to 6799.87 Å). The most complete tables were published, in 1911, by Exner and Haschek, who list 905 lines (2320.95 to 6799.91 Å) observed in the arc spectrum [8] and 795 lines (2224.58 to 6489.31 Å) in the spark spectrum [9]. These tables contain a considerable number of lutecium and thulium lines, and many ytterbium lines are common to both, without positive indication as to whether they belong to neutral or to ionized atoms. In 1914, Blumenfeld and Urbain [10] measured the wave lengths of 499 lines (2309.6 to 3499.4 Å) most of which were believed to represent ytterbium, although many were recognized as possible impurities. The latest extensive description of Yb spectra consists of 422 lines (2271.51 to 7699.49 Å) observed in the arc spectrum by Eder [11], neglecting more than 200 lines ascribed to impurities, principally Tm. The salts available for these early investigations of rare-earth spectra were so impure that great difficulty was experienced in sorting the lines and frequently the same lines were ascribed to two different elements. For example, [12] 2615.41 Yb, Lu; 2693.33 Yb, Lu; 3283.40 Yb, Tm; 3362.61 Yb, Tm; 3425.10 Yb, Tm; 3441.51 Yb, Tm; 3472.48 Yb, Lu; 3887.34 Yb, Er; 5074.32, De, Tm, Yb; 5307.11 Yb, Tm. Except for the temperature classification of 399 lines (2935.09 to 6799.58 Å) by King [13], no further investigations of Yb spectra have been reported in the past 22 years.

## II. EXPERIMENTS

Our first measurements of Yb spectra were made in 1929 when the investigation of Lu spectra was begun [4]. Samples of Lu and Yb oxides prepared by Auer, and obtained from Eder in 1919 when he discontinued his work in spectroscopy, were available. Although the chemical separation was far from complete, it was possible to make a fairly satisfactory assignment of lines to Lu and Yb by simultaneous comparison of arc and spark spectra. However, both salts contained some Tm as impurity which could not be completely identified without comparable data for Tm spectra. The first sample of Tm salt available for this purpose was generously supplied in 1933, by Professor B. S. Hopkins, University of Illinois, who also supplied a sample of Lu oxide and three samples of Yb oxide, one of which contained more Lu than Tm, another more Tm than Lu, and the third was entirely free from both Lu and Tm, but contained considerable lanthanum. This ytterbium material was purified by the electrolytic reduction method devised by L. F. Yntema [14]. Alkali (Li, Na, K) and alkaline-earth (Mg, Ca, Sr, Ba) impurities are usually present in rare-earth salts, but these have extremely simple spectra, which are

easily identified. Miscellaneous impurities were detected by comparing our final wave-length list with Kayser's Hauptlinien [12] and with a description of lanthanum spectra [15].

The procedures employed to insure the correct assignment of spectral lines to their emitter, both as regards chemical element and stage of ionization, were the same as for the study of Lu spectra [4], and the details regarding light sources, spectrographs, photographic plates, etc., given there need not be repeated here.

### III. RESULTS

The measured wave lengths and estimated relative intensities of 1,668 Yb lines observed in arc and spark spectra are presented in table 1. Each line in this table was observed on two or more spectrograms except a few with questioned intensities which were seen only on the strongest exposures. The brightest Yb lines were measured on eight or nine spectrograms, and many of them were measured a number of times on Lu and Tm spectrograms, where they appeared as impurities. The concordance of values from different spectrograms and the coincidence of wave lengths with impurity lines indicates that the final results for most lines are correct within 0.01 Å, but errors of 0.02 Å or more may exist among faint lines. Hazy lines which occur so frequently in spark spectra are not only subject to larger accidental errors but may show some systematic error, since they are usually unsymmetrical. Otherwise, Yb lines appear to be sharp and free from hyperfine structure. No information on the isotopic constitution of Yb is available, but since hyperfine structures have been detected only for lines of rare earths with odd atomic number, Yb ( $Z=70$ ) may be assumed to be immune.

In 1915 Eder [11] published 177 lines, which he suspected belonged to a new element, denebium (De). It now appears that at least 110 of these lines are due to ytterbium, and many of the remainder are due to thulium.

TABLE 1.—Arc and spark spectra of ytterbium ( $Z=70$ )

$d$ =Double.  
 $e$ =Enhanced at electrode.  
 $h$ =Hazy.  
 $H$ =Very hazy.

I=Shaded to longer waves.  
I=Spectrum of neutral atoms.  
II=Spectrum of singly ionized atoms.

III=Spectrum of doubly ionized atoms.

$\lambda$ air Å	Intensity and character		Spectrum	$\lambda$ air Å	Intensity and character		Spectrum	$\lambda$ air Å	Intensity and character		Spectrum
	Arc	Spark			Arc	Spark			Arc	Spark	
2073.70	-----	8h	II	2102.72	20	200	II	2127.54	-----	2h	II
2078.12	-----	10h	II	2106.46	-----	2	II	2128.82	-----	2	II
2083.65	-----	2	II	2108.20	-----	5	II	2129.64	-----	4h	II
2087.37	-----	3h	II	2109.60	-----	100h	II	2130.45	-----	2h	II
2088.05	-----	15h	II	2110.26	-----	5h	II	2131.37	10	20	II
2091.28	-----	20h	II	2110.60	-----	3	II	2131.71	-----	2h	II
2092.31	-----	30h	II	2115.46	-----	4h	II	2133.17	-----	3	II
2093.18	-----	6h	II	2115.84	-----	2	II	2133.34	-----	3	II
2094.82	-----	10h	II	2116.65	50	250	II	2135.66	-----	4h	II
2095.38	-----	200h	III?	2117.83	-----	4h	II	2134.98	-----	3h	II
2096.84	-----	30h	II	2119.25	-----	30h	II	2135.22	-----	20h	II
2097.41	-----	2	II	2121.54	-----	15	II	2136.33	-----	2	II
2098.16	-----	3	II	2122.81	-----	3	II	2137.52	-----	9	II
2098.40	-----	50h	II	2123.30	-----	15	II	2137.71	-----	15 Lu?	II
2102.10	-----	3h	II	2126.72	40	200	II	2138.32	-----	10	II

TABLE 1.—Arc and spark spectra of ytterbium (Z=70)—Continued

$\lambda$ air A	Intensity and character		Spec- trum	$\lambda$ air A	Intensity and character		Spec- trum	$\lambda$ air A	Intensity and character		Spec- trum
	Arc	Spark			Arc	Spark			Arc	Spark	
2139.25	-----	2	II	2211.87	-----	2	II	2373.89	2	8	II
2139.96	-----	25	II	2214.64	-----	8	II	2377.32	-----	2h	II
2140.41	-----	2	II	2221.47	-----	2h?	II	2380.37	2	9	II
2141.00	-----	10	II	2224.45	20	40	II	2382.56	-----	2	II
2141.67	-----	5	II	2224.83	-----	2	II	2385.00	1	5	II
2142.18	-----	4	II	2227.70	-----	3	II	2388.40	1	3	II
2143.42	-----	5?	II	2231.65	-----	2	II	2390.72	10	30	II
2143.86	-----	4?	II	2237.48	-----	3	II	2397.32	-----	1?	II
2144.74	-----	30 Ag?	II	2240.09	5	25	II	2397.83	-----	3	II
2146.46	-----	3?	II	2244.25	-----	8h	II	2398.00	10	5	{ I, II?
2147.15	-----	1	II	2255.46	-----	1?	II	2403.39	-----	3	II
2147.50	-----	3h	II	2257.01	4	40	II	2404.07	-----	4h	II
2148.07	-----	8	II	2261.12	-----	3	II	2406.07	1	5	II
2148.49	-----	4	II	2262.25	2	30	II	2418.39	-----	2h	II
2148.92	-----	10 Cu?	II	2263.16	-----	3h	II	2421.86	8	25	II
2149.43	-----	3	II	2263.85	-----	2	II	2422.81	2	4	II
2150.45	-----	3	II	2265.65	3	50	II	2424.56	-----	2h	II
2150.93	-----	2	II	2267.15	-----	1	II	2425.65	-----	2h	II
2151.20	-----	2h	II	2268.28	1	4	II	2426.16	-----	1	II
2152.32	3?	10h	II	2268.64	-----	3h	II	2433.62	-----	6h	II
2153.44	-----	1	II	2271.12	8 ?	-----	I	2439.30	-----	4	II
2153.67	-----	1	II	2276.07	2 ?	-----	I	2439.67	-----	3	II
2154.16	-----	80h	{ II, III?	2282.99	5	50	II	2440.57	-----	5h	II
2155.18	-----	7	II	2283.38	8	10	II	2447.23	-----	4	II
2155.51	-----	40	II	2283.98	-----	8h	II	2454.7	-----	20II	II
2156.51	-----	3	II	2285.80	3 ?	-----	I	2456.00	-----	5h	II
2156.79	-----	2	II	2285.85	-----	8h	II	2457.65	-----	2	II
2157.87	-----	3h	II	2288.93	1	6	II	2458.23	-----	1h	II
2158.79	-----	4	II	2291.63	-----	3h	II	2460.24	6	30 Ag?	II
2159.29	-----	1	II	2292.27	-----	3h	II	2461.38	-----	5h	II
2159.88	-----	3h	II	2292.81	1	3	II	2464.48	20	5	I
2160.27	-----	10h	II	2297.88	2	4	II	2465.11	-----	3?	II
2161.60	100	250	II	2298.65	1	4	II	2466.62	2	10	II
2163.49	-----	1	II	2303.30	-----	2d Lu?	II	2471.04	-----	2?	II
2163.87	-----	10	II	2305.33	8	100	II	2481.40	1	4	II
2165.19	-----	8h	II	2307.41	1	6h	II	2484.28	-----	2	II
2165.53	-----	3	II	2308.49	-----	2h	II	2484.88	3	8	II
2168.09	-----	3	II	2309.26	-----	20	II	2487.03	-----	2	II
2169.10	-----	7	II	2310.67	2? Lu?	-----	I	2488.96	-----	3	II
2169.77	-----	5	II	2312.56	1	2	II	2490.45	2	40	II
2172.13	-----	3	II	2314.48	4	50	II	2491.69	-----	3	II
2172.58	-----	2	II	2315.20	5	10	II	2493.63	2	5	II
2173.34	2	5	II	2320.80	20	5	I	2495.03	-----	1	II
2174.28	4	10	II	2323.17	1	2	II	2495.69	-----	10h	II
2175.38	-----	3 Lu?	II	2326.91	2	6h	II	2498.35	1	4	II
2175.88	-----	1	II	2333.39	-----	2	II	2500.57	-----	2	II
2177.04	-----	1	II	2335.42	3 ?	-----	I	2500.89	-----	1	II
2177.50	-----	5	II	2335.47	-----	15h	II	2501.17	-----	2	II
2178.80	-----	8h	II	2337.95	2	40	II	2501.99	10	30	II
2180.24	-----	7h	II	2340.39	2?	-----	I	2505.46	5	10	II
2181.51	-----	2	II	2340.46	-----	8h	II	2508.04	2?	2	II
2181.85	-----	2	II	2340.75	-----	3h	II	2509.85	-----	2	II
2182.55	-----	3h	II	2344.65	5	20	II	2510.27	-----	1	II
2183.29	-----	7	II	2349.39	1	8	II	2510.50	1	5	II
2183.61	-----	3h	II	2350.03	-----	1	II	2512.04	20	100	II
2184.78	-----	2 Lu?	II	2361.09	-----	6	II	2512.58	-----	10h	II
2185.69	60	100	II	2362.87	10	20	II	2515.60	-----	5	II
2188.25	-----	5	II	2363.48	-----	4h	II	2516.36	4	10	II
2189.42	-----	8	II	2365.45	-----	15	II	2516.83	3	30	II
2193.08	-----	2	II	2366.78	3	12	II	2517.99	-----	7	II
2193.40	-----	4d	II	2367.58	-----	10h	II	2520.32	-----	3	II
2195.16	-----	2	II	2369.44	-----	10	II	2521.02	1	6	II
2198.14	-----	20h	II	2369.87	-----	1?	II	2522.41	8	15	II
2201.19	-----	3?	II	2370.10	-----	3h	II	2524.98	-----	4	II
2203.07	-----	2?	II	2373.06	1	5	II	2526.28	-----	5h	II

TABLE 1.—Arc and spark spectra of ytterbium (Z=70)—Continued

$\lambda$ air A	Intensity and character		Spectrum	$\lambda$ air A	Intensity and character		Spectrum	$\lambda$ air A	Intensity and character		Spectrum
	Arc	Spark			Arc	Spark			Arc	Spark	
2527.85	2	10	II	2619.90	2	5	II	2717.66	-----	3	II
2528.32	-----	7	II	2621.13	2e	100	III	2718.35	7	40	II
2529.54	-----	4h	II	2621.66	1	4	II	2719.00	-----	2?	II
2532.58	-----	3h	II	2623.20	1	5	II	2722.19	-----	6	II
2536.01	1	10	II	2627.04	-----	20	II	2729.06	-----	1	II
2537.64	8	25	II	2628.04	1	4	II	2729.49	-----	2	II
2538.18	2	6	II	2628.78	-----	2	II	2731.53	-----	1	II
2538.68	30	40	II	2631.70	1	4	II	2732.74	8	40	II
2539.63	-----	2	II	2632.66	-----	4	II	2734.08	1	6	II
2540.61	-----	2	II	2634.32	5	20	II	2736.56	-----	3	II
2541.1	-----	3H	II	2635.39	-----	5h	II	2737.61	-----	3	II
2542.82	-----	8H	II	2636.49	-----	5h	II	2738.81	-----	2	II
2545.85	2	2	II	2638.09	1e	60	III	2740.38	-----	1	II
2547.30	-----	1	II	2639.44	5	15	II	2740.78	-----	1	II
2547.47	-----	4	II	2640.52	-----	15h	II	2741.15	-----	1	II
2548.06	-----	1h	II	2641.90	4	10	II	2741.72	2	20	II
2548.73	-----	8	II	2642.55	5e	150	III	2742.64	-----	1	II
2550.04	2	4	II	2643.65	-----	5	II	2745.11	-----	2	II
2550.42	-----	3	II	2644.31	5	40	II	2745.70	-----	4	II
2550.79	-----	4h	II	2646.45	2	10	II	2747.59	2	20	II
2552.13	10	40	II	2647.25	-----	4	II	2748.03	-----	3	II
2552.69	8	30	II	2647.46	2	7	II	2748.65	10	40	II
2555.31	-----	50h	II	2648.80	2	8	II	2749.64	-----	2	II
2556.26	-----	3	II	2649.78	4	10	II	2749.96	-----	10	II
2557.24	2	8	II	2650.73	2	4	II	2750.47	50	200	II
2557.70	-----	4	II	2651.72	2e	60	III	2751.44	3	20	II
2559.03	-----	1	II	2652.23	2e	60	II	2753.67	-----	3h	II
2559.94	-----	2	II	2653.74	50	200	II	2754.93	-----	4	II
2560.57	-----	5	II	2656.11	2	7	II	2756.02	-----	4	II
2561.64	-----	1	II	2659.27	2	5	II	2756.79	-----	3?	II
2565.56	5	15	II	2660.01	-----	6	II	2758.99	1	10	II
2566.80	-----	3	II	2661.86	-----	1	II	2759.54	-----	4	II
2567.63	10e	300	III	2665.02	10	60	II	2760.78	3	25	II
2568.17	-----	2h	II	2666.11	5e	150	III	2761.38	3	25	II
2571.34	6	20	II	2666.98	5e	150	III	2764.42	1	10	II
2572.11	-----	3	II	2668.75	2	20	II	2765.04	-----	1	II
2573.13	3	10	II	2671.98	10	2	I	2765.54	-----	7	II
2574.78	1	3	II	2672.65	20	80	II	2768.27	-----	2	II
2577.68	2	30	II	2673.37	-----	2	II	2771.34	3	30	II
2579.58	5e	200	III	2674.87	-----	4	II	2774.32	-----	4	II
2583.72	-----	2	II	2676.13	-----	2	II	2775.45	-----	2	II
2586.28	1	4	II	2677.37	-----	40	II	2776.28	10	50	II
2588.65	-----	8h	II	2678.60	-----	1?	II	2780.03	-----	3	II
2591.02	1	5	II	2680.40	-----	3	II	2782.21	-----	1	II
2592.71	-----	3	II	2683.42	-----	10	II	2782.57	-----	1	II
2594.18	1	5	II	2684.75	3	25	II	2784.66	3	30	II
2594.48	2	5	II	2685.99	-----	3	II	2785.80	-----	2	II
2596.15	2	6	II	2687.95	-----	5	II	2787.00	-----	1	II
2596.29	2	7	II	2690.99	-----	40	II	2787.98	-----	4	II
2596.73	2	3	II	2691.99	-----	6	II	2788.31	-----	10	II
2597.26	1	20	II	2692.40	-----	2	II	2789.44	-----	3	II
2599.16	2	50	II	2692.69	-----	2	II	2791.58	-----	1	II
2600.19	1	3	II	2694.61	-----	1?	II	2793.25	2	10	II
2600.84	1	3	II	2695.43	1	9	II	2794.44	1	7	II
2602.41	-----	1	II	2696.63	-----	6	II	2794.78	-----	4	II
2604.04	4	10	II	2700.80	-----	10	II	2795.10	-----	4	II
2607.86	3	7	II	2704.52	-----	3	II	2795.62	-----	10	II
2608.45	-----	2	II	2708.08	-----	1	II	2797.79	-----	8	II
2609.12	-----	4	II	2708.84	1	5	II	2798.21	-----	10	II
2610.86	3	20	II	2709.72	-----	2	II	2799.39	2	3	II
2612.03	1	4	II	2710.56	3	15d	II	2800.04	3	15	II
2612.60	1	4	II	2712.66	3	7	II	2803.44	-----	80	II
2615.25	5	20	II	2714.42	-----	5	II	2804.27	-----	4	II
2617.00	6	25	II	2715.94	-----	2	II	2806.05	-----	1	II?
2619.06	2	6	II	2717.19	-----	2h	II	2807.22	-----	3d	II

TABLE 1.—Arc and spark spectra of ytterbium (Z=70)—Continued

$\lambda$ air A	Intensity and character		Spec- trum	$\lambda$ air A	Intensity and character		Spec- trum	$\lambda$ air A	Intensity and character		Spec- trum
	Arc	Spark			Arc	Spark			Arc	Spark	
2807.82	-----	1	II	2881.92	-----	5	II	2970.84	2	20	II
2808.30	-----	5	II	2882.14	-----	4	II	2972.50	-----	3	II
2808.56	-----	2	II	2883.87	-----	2h	II	2975.58	-----	2	II
2809.33	-----	1	II	2884.42	1?	-----	I	2977.54	-----	4	II
2810.11	-----	3	II	2885.07	-----	1	II	2978.91	-----	3	II
2810.72	-----	4	II	2885.98	2	7	II	2979.69	-----	2	II
2811.17	-----	1?	II	2886.26	3	10	II	2979.86	-----	4	II
2812.66	-----	1?	II	2888.03	8	40	II	2981.52	-----	2	II
2814.25	-----	3	II	2891.38	100	200	II	2982.09	-----	2	II
2814.53	2	10	II	2893.63	2	15	II	2982.50	3	15	II
2814.87	-----	1	II	2894.99	-----	10d	II	2982.65	1	4	II
2816.34	1	4	II	2896.91	1	10	II	2983.70	2	8	II
2816.96	-----	40	II	2898.34	-----	20	II	2983.98	10	70	II
2818.75	1	150	{ II?	2899.24	-----	2	II	2984.81	-----	5?	II
2819.47	-----	2	II	2899.72	4	20	II	2985.08	5	30	II
2821.15	10	50	II	2902.39	-----	2	II	2985.87	2	9	II
2823.57	-----	6	II	2902.92	-----	6	II	2987.93	-----	2	II
2824.20	-----	2	II	2906.34	-----	40	II	2989.21	-----	2	II
2824.97	3	20	II	2906.88	-----	4	II	2989.79	-----	1	II
2827.90	-----	4	II	2908.10	-----	4	II	2990.36	3	15	II
2828.49	-----	2	II	2908.33	1	5	II	2991.86	4	20	II
2830.98	8	80	II	2909.19	2	7	II	2993.93	1	7	II
2832.20	-----	4	II	2909.47	2	8	II	2994.81	10	80	II
2834.98	1	15	II	2911.51	5	40	II	2995.86	2	10	II
2838.65	-----	3	II	2912.86	-----	3	II	2998.04	-----	9	II
2841.34	-----	4	II	2914.21	10	60	II	2998.38	-----	3	II
2842.19	-----	1	II	2914.49	-----	4?	II	3000.46	5	40	II
2842.30	-----	1	II	2915.27	10	40	II	3001.29	-----	3	II
2842.58	-----	2	II	2916.43	1	10	II	3002.04	-----	3	II
2843.01	-----	4	II	2916.84	-----	1	II	3002.61	3	20	II
2843.80	-----	10	II	2919.34	15	90	II	3005.76	30	200	II
2843.80	-----	3h	II	2921.12	4	20	II	3006.86	1	4	II
2844.77	-----	3h	II	2924.23	3	15	II	3009.39	6	40	II
2846.29	-----	1h	II	2927.12	-----	1	II	3010.61	5	30	II
2847.18	10	?	I?	2927.86	2	10	II	3014.45	4	30	II
2847.25	?	60h	II	2934.34	3	? Ag	I, II?	3017.56	10	80	II
2848.44	4	30	II	2935.10	3	15	II	3019.06	-----	3	II
2849.33	1	8	II	2937.18	2	10	II	3019.47	-----	2	II
2851.12	20	100	II	2938.19	-----	3	II	3020.70	-----	10	II
2851.86	-----	4	II	2939.52	2	6	II	3022.45	1	7	II
2853.40	1	8	II	2940.51	3	25	II	3023.62	-----	4	II
2853.68	-----	2	II	2942.04	1	6	II	3024.93	-----	1	II
2854.12	3	20	II	2942.82	-----	5	II	3026.67	10	80	II
2854.49	2	15	II	2944.47	-----	3	II	3028.39	-----	2	II
2855.86	-----	2	II	2945.91	10	60	II	3029.53	-----	70h	{ II?
2856.95	-----	2?	II	2946.30	3	15	II	3031.11	50	60	II
2858.35	2	10	II	2946.77	1	8	II	3031.62	-----	2?	II
2858.45	1	8	II	2947.11	-----	2	II	3033.85	-----	4	II
2859.39	4	40	II	2950.32	2	10	II	3034.64	3	10	II
2859.80	20	60	II	2951.02	-----	4	II	3036.83	-----	4	II
2860.40	2	15	II	2951.41	-----	2	II	3038.00	2	6	II
2861.21	6	60	II	2953.03	-----	5	II	3038.55	-----	2	II
2861.31	5	50	II	2955.31	3	9	II	3039.67	4	25	II
2862.87	-----	1	II	2957.64	-----	3	II	3040.49	1	5	II
2863.19	-----	2h	II	2960.85	1	5	II	3042.65	4	30	II
2864.78	-----	3d	II	2961.81	-----	3	II	3044.02	1	5	II
2865.34	-----	2	II	2962.53	4	20	II	3044.83	1	5	II
2866.18	-----	5	II	2963.21	-----	10	II	3046.48	3	15	II
2867.05	10	80	II	2963.46	2	20	II	3047.06	2	9	II
2868.55	-----	1	II	2964.41	-----	5h	II	3053.98	-----	2	II
2870.06	2	20	II	2964.75	8	30	II	3054.67	-----	1	II
2871.71	-----	1	II	2965.16	-----	4	II	3055.16	1	5	II
2872.15	-----	2	II	2966.76	-----	15	II	3063.14	2	7	II
2873.50	1?	-----	I	2969.60	-----	1?	II	3063.69	-----	8	II
2875.90	-----	15	II	2970.56	100	150	II	3064.92	-----	2	II
2879.16	1	10	II								

TABLE 1.—Arc and spark spectra of ytterbium (Z=70)—Continued

$\lambda$ air A	Intensity and character		Spec- trum	$\lambda$ air A	Intensity and character		Spec- trum	$\lambda$ air A	Intensity and character		Spec- trum
	Arc	Spark			Arc	Spark			Arc	Spark	
3065.04	10	50	II	3175.75	1	7	II	3315.38	-----	5	II
3067.38	-----	4	II	3180.91	4	40	II	3316.83	-----	3	II
3068.29	-----	4	II	3186.59	-----	4	II	3319.18	2	15	II
3071.61	-----	4	II	3190.81	-----	1	II	3319.43	4	-----	I
3073.70	1	5	II	3191.41	-----	10h	II	3320.31	-----	4	II
3074.52	-----	1	II	3192.88	20	120	II	3322.97	-----	5	II
3076.04	1	4	II	3194.24	1	4	II	3324.16	1	5	II
3080.15	-----	1?	II	3194.76	-----	5	II	3324.46	-----	1	II
3080.55	-----	1	II	3195.57	2	9	II	3325.57	-----	4	II
3084.36	-----	6	II	3196.06	-----	2	II	3327.77	-----	2	II
3085.82	-----	6	II	3196.35	-----	2	II	3331.21	-----	1	II
3086.99	1	6	II	3198.66	4	30	II	3333.07	3	15	II
3087.99	-----	2?	II	3199.81	-----	2	II	3337.18	6	2	I
3089.11	5	20	II	3201.16	8	60	II	3342.94	8	2	I
3089.61	-----	2?	II	3204.69	1	4	II	3343.06	2?	10	II
3090.79	-----	2	II	3206.16	1	4	II	3346.50	3	-----	I
3092.54	-----	40	II	3207.70	2	2	II	3347.63	3	15	II
3093.44	-----	3	II	3210.11	1	10	II	3349.94	-----	1	II
3093.87	5	25	II	3214.33	-----	1	II	3351.08	2	1	I
3094.92	-----	3	II	3215.49	-----	2	II	3352.49	4	-----	I
3095.24	-----	2	II	3215.94	-----	2	II	3353.74	-----	2	II
3097.34	-----	1?	II	3217.19	4	20	II	3355.88	-----	4	II
3101.36	2	20	II	3218.33	3	15	II	3356.98	-----	4	II
3102.07	1	10	II	3221.26	-----	4	II	3361.61	-----	1	II
3107.79	-----	30?	II	3221.51	-----	3	II	3362.43	3	-----	I
3107.91	15	100	II	3223.17	-----	1?	II	3363.61	2	-----	I
3109.79	-----	2	II	3225.86	4	20	II	3365.97	-----	7	II
3112.98	-----	2	II	3226.75	-----	3	II	3369.90	-----	1?	II
3115.33	4	30	II	3228.63	-----	15	II	3375.48	5	25	II
3116.07	1	4	II	3229.84	-----	3	II	3378.43	-----	1	II
3116.50	-----	5	II	3231.99	2	9	II	3379.78	-----	5	II
3116.71	3	9	II	3236.16	3	15	II	3382.58	2?	-----	I
3117.81	15	50	II	3237.82	1?	-----	I	3384.04	-----	5	II
3122.22	-----	2	II	3239.20	2	10	II	3387.49	5	1	I
3123.52	-----	3	II	3239.60	2	-----	I	3390.32	3	-----	I
3125.45	1	2	II	3242.07	-----	2	II	3391.10	1	9	II
3126.06	-----	40h	II	3244.97	-----	2h?	II	3392.40	-----	1	II
3127.14	-----	2	II	3246.07	2	-----	I	3394.44	1	5	II
3127.86	1	6	II	3252.14	-----	1?	II	3396.32	-----	4	II
3129.14	-----	3	II	3254.22	1	3	II	3400.58	1	-----	I
3132.61	2	8	II	3256.05	-----	1	II	3401.02	2	7	II
3135.14	-----	2	II	3259.10	2	10	II	3402.28	-----	1	II
3136.76	3	20	II	3261.51	4	25	II	3404.11	2	7	II
3138.63	-----	6	II	3261.70	-----	7	II	3408.50	-----	2	II
3140.92	10	40	II	3264.92	-----	2	II	3409.88	-----	1?	II
3141.72	3	15	II	3265.08	-----	2	II	3410.54	-----	1	II
3145.06	4	20	II	3265.95	1?	-----	II	3411.24	1	-----	I
3145.55	3	2	I	3271.18	-----	4	II	3412.45	2	1	I
3146.13	-----	2	II	3271.53	-----	3	II	3414.86	-----	2	II
3148.99	1	5	II	3275.81	1	6	II	3415.04	-----	1?	II
3150.44	-----	1?	II	3286.97	-----	5	II	3415.98	-----	2	II
3151.45	-----	2	II	3289.36	400	800	II	3416.88	-----	3	II
3152.45	-----	2	II	3294.33	-----	4	II	3418.40	3	1	I
3153.17	2	7	II	3297.85	-----	4	II	3419.62	-----	3	II
3153.86	4	40	II	3298.82	-----	1	II	3420.35	1	-----	I
3155.18	3	20	II	3302.42	-----	1	II	3426.06	10	3	I
3155.79	-----	2	II	3304.57	2	10	II	3428.48	4	20	II
3158.29	2	25	II	3304.76	2	10	II	3431.14	8	3	I
3162.31	2	-----	I	3305.27	3	-----	I	3434.62	2	5	II
3163.79	3	25	II	3305.73	5	30	II	3436.46	2	7	II
3165.20	2	20	II	3306.79	1	7	II	3438.73	3	20	II
3168.41	-----	2	II	3309.38	2	8	II	3438.83	4	30	II
3169.05	3	30	II	3310.95	-----	1	II	3443.59	1	-----	I
3171.19	1	5	II	3313.50	-----	1?	II	3446.89	2	6	II
3173.77	1	8	II	3315.11	2	-----	I	3448.00	-----	1	II

TABLE 1.—Arc and spark spectra of ytterbium (Z=70)—Continued

$\lambda$ air A	Intensity and character		Spec- trum	$\lambda$ air A	Intensity and character		Spec- trum	$\lambda$ air A	Intensity and character		Spec- trum
	Arc	Spark			Arc	Spark			Arc	Spark	
3452.41	3	-----	I	3634.54	3	—	I	3887.31	—	8	II
3454.08	10	60	II	3637.76	7	15	II	3890.86	—	1?	II
3456.30	-----	1?	II	3644.23	1	3	II	3900.86	20	2	I
3458.27	4	20	II	3648.13	2	—	I	3904.82	3	20	II
3458.36	4	-----	I	3648.46	—	2	II	3905.88	—	2	II
3460.27	7	2	I	3649.74	—	2	II	3908.02	2?	—	I
3461.71	1	6	II	3652.96	—	1?	II	3911.28	6	—	I
3462.36	1?	-----	I	3655.73	3	—	I	3934.30	—	4	II
3464.36	100	20	I	3664.56	—	2	II	3938.26	1	5	II
3467.05	-----	3	II	3664.76	—	2	II	3938.53	1	4	II
3469.92	-----	2	II	3669.71	6	10	II	3946.94	—	3	II
3470.78	-----	2	II	3670.68	1	8	II	3948.23	—	1?	II
3474.81	2	5	II	3675.08	5	20	II	3949.59	1	1	I
3476.31	30	8	I	3687.15	—	2	II	3949.88	—	1?	II
3478.84	15	80	II	3687.60	—	3	II	3968.03	—	3	II
3485.76	3	10	II	3690.54	—	10?	II	3975.28	3	—	I
3488.42	3	-----	I	3691.48	—	4	II	3980.37	1?	—	I
3488.80	1	3	II	3691.71	2	—	I	3987.98	2000	100	I
3491.63	-----	1?	II	3694.19	500	1000	II	3990.89	40	4	I
3495.92	5h	-----	I	3698.59	3	15	II	3993.75	2	—	I
3502.18	-----	3	II	3699.50	2	—	I	3994.86	1	-----	I
3506.14	-----	3	II	3699.81	—	4	II	3999.81	-----	2	II
3507.83	3	15	II	3700.57	3	—	I	4000.89	-----	6h	II
3515.86	1	4	II	3701.03	2?	—	I	4007.36	5	-----	I
3517.02	1?	-----	I	3703.42	—	4	II	4009.24	-----	1?	II
3518.16	1	5	II	3708.66	1	3	II	4019.35	-----	7	II
3520.28	5	30	II	3710.32	2	7	II	4024.01	-----	2	II
3531.24	-----	2	II	3716.15	2?	—	I	4028.27	-----	8h	II
3534.07	-----	1?	II	3720.98	—	2	II	4032.92	-----	1	II
3539.33	-----	2	II	3722.29	2	10	II	4040.09	1	8	II
3541.40	-----	1	II	3724.22	3	15	II	4043.06	-----	8h	II
3542.40	-----	2?	II	3729.80	—	2	II	4047.39	-----	4	II
3543.15	-----	2	II	3730.39	1	3	II	4050.10	-----	3	II
3545.71	2	-----	I	3734.70	6	1	I	4052.28	8	2	I
3549.82	4	15	II	3741.61	—	3	II	4056.18	-----	10	II
3552.33	1	3	II	3743.15	2?	—	I	4074.70	-----	2	II
3559.00	1	-----	I	3749.68	2	7	II	4077.27	4	30	II
3560.33	6	15	II	3753.05	—	2?	II	4082.98	2	-----	I
3560.71	4	30	II	3753.89	—	3	II	4086.62	-----	1	II
3562.70	-----	1	II	3761.00	—	4	II	4089.69	40	7	I
3563.93	1	5	II	3762.54	1	6	II	4091.49	1	4	II
3567.12	1	7	II	3766.10	1	4	II	4093.73	1?	-----	I
3570.56	3	10	II	3770.10	15	3	I	4097.87	2	8	II
3572.49	-----	2	II	3773.43	2	3	II	4109.60	5h	-----	I
3574.54	2h	-----	I	3774.32	2	—	I	4113.04	2	7	II
3575.58	-----	1?	II	3779.28	—	2	II	4119.26	4	-----	I
3577.02	1	5	II	3782.54	2	15	II	4119.46	5	10?	II
3582.83	-----	1?	II	3784.40	—	3	II	4122.86	1	6	II
3585.47	15	25	II	3786.37	2?	—	I	4127.34	2	-----	I
3586.84	-----	2?	II	3791.74	8h	2	I	4131.00	-----	3h?	II
3600.38	-----	1	II	3798.18	2	—	I	4132.14	3h	-----	I
3600.74	-----	4	II	3798.44	2?	—	I	4135.10	10	60	II
3603.85	-----	3	II	3807.54	3	15	II	4139.05	2	-----	II
3606.47	5	25	II	3814.22	—	8	II	4141.49	-----	2	II
3608.48	-----	3	II	3816.21	2	7	II	4147.80	-----	2	II
3610.23	1	4	II	3816.34	—	2?	II	4149.07	40	6	I
3611.30	2	8	II	3839.46	1	3	II	4164.64	-----	2	II
3614.05	-----	2	II	3839.92	5	—	I	4169.12	1?	-----	I
3618.08	-----	2	II	3840.33	—	3	II	4170.11	3	20	II
3619.81	8	30	II	3847.48	—	3	II	4172.24	-----	2	II
3621.00	1	5	II	3847.86	3	—	I	4172.36	1?	-----	I
3624.65	-----	2	II	3857.61	3	—	I	4174.57	10	1	II
3629.91	1	3	II	3858.55	—	2	II	4180.82	40	100	II
3631.08	-----	2	II	3863.46	1	5	II	4186.90	-----	10h	II
3632.56	-----	3d	II	3872.85	20	2	I	4190.30	4	30	II

TABLE 1.—Arc and spark spectra of ytterbium (Z=70)—Continued

$\lambda$ air A	Intensity and character		Spec- trum	$\lambda$ air A	Intensity and character		Spec- trum	$\lambda$ air A	Intensity and character		Spec- trum
	Arc	Spark			Arc	Spark			Arc	Spark	
4210.30	4		I	4472.43	20h		I	4743.38	1?		I
4216.72		5	II	4480.27		2	II	4745.52		2	II
4218.55	4	50	II	4482.44	10	2	I	4746.70		15h	II
4218.64	15	?	I	4484.43		3h	II	4750.01		4h	II
4224.19		1	II	4487.27	2	10	II	4751.77	4		I
4227.94		10	II	4488.28	4		I	4752.93	3	20h	II
4229.66	2		I	4493.96	2	10	II	4758.32	2	1	I
4230.24		4	II	4503.51		2h?	II	4761.16		5h	II
4231.99	10	2	I	4503.62	5		I	4765.37	3h		I
4233.45	2?		I	4513.40	6	1	I	4770.17	2		I
4234.55	2	15	II	4514.04		2?	II	4774.61	3h		I
4244.21		2h	I	4515.16	30	100	II	4774.84	3h		I
4247.88	2	7	II	4520.66		1?	I	4779.00	3		I
4251.52	4	1	I	4529.90	20	2	I	4781.89	200	5	I
4252.53	5	20	II	4533.50	4	1	I	4784.52	3		I
4254.78	2	4	II	4536.80		2?	II	4785.34	2		I
4255.78	3	8	II	4539.30		2	II	4786.62	100	500	II
4256.77	3h		I	4539.70	1?		I	4797.85	1		I
4257.66	3	15	II	4541.35		6	I	4808.51		4h	I
4266.99	2	8h	II	4543.90		4	II	4809.47		2	II
4270.55		3h	II	4547.01		1	II	4812.91		3h	I
4272.12	2		I, II?	4548.35	2	2	I, II?	4816.40	40h	1	I
4273.32		3	II	4553.58	10	40	II	4818.38	3	30	II
4277.73	9h	2	I	4557.94	2	6	I	4820.25	20	80	II
4283.01	7 Ca?		I	4564.00	50	2	I	4830.70	2		I
4300.99	6	1	I	4565.99		2h?	II	4831.31	2		I
4305.49	2		I	4567.38	5	1	I	4831.55		2h	II
4305.96	20h	3	I	4568.89	2h		I	4931.96	3h		I
4306.49		3	II	4576.21	200	10	I	4834.72		10h	II
4309.82	8		I	4580.75	2		I	4836.96	20	120	II
4312.36	2		I	4582.36	50	5	I	4837.47	20	1	I
4316.96	10	40	II	4585.82		2h	II	4841.18	2	2	I
4322.23	3	20	II	4587.14		2h	II	4848.45	3	20h	II
4325.28		3	I	4589.22	20	1	I	4851.17	8	10	II
4326.40	5h		I	4590.84	30	?	I	4853.84	3		I
4329.72	1		I	4593.36	2		I	4871.16		8h	II
4336.48		3h?	II	4597.25		2	II	4871.64		3h	II
4339.10	2	10	II	4598.37	10	50	II	4893.48	2		I
4344.21	2		I	4610.22	2h		II	4894.63	50		I
4344.29		2?	II	4610.66		4?	II	4894.98	3	10	II
4352.94	6h		I	4618.53		3?	II	4897.94		2	II
4356.65	3		I	4624.46	4h		I	4903.72		8h	II
4363.30		6	II	4633.18	2		I	4912.38	20		I
4370.81	5	40	II	4634.04	2	15h	I	4917.04		4h	II
4371.42	1?		I	4641.62	2		I	4918.14	2?		I
4372.60		1	II	4644.54	10	1	I	4918.46		3h?	II
4376.48	4		I	4650.06	10		I	4929.31		3h	II
4376.92		1h	II	4651.67	3		I	4931.95	10		I
4386.10		2h?	I	4657.04	9h		I	4935.51	500	20	I
4389.76	1	10	II	4658.47	1?		I	4936.96	8h		I
4392.83	2	15	II	4661.81	2h		I	4937.22	5	100	II
4393.75	30h		I	4667.08	2		I	4942.23	1?		I
4396.49	1	2	I	4670.59	3	7	II	4942.78	1	5h	II
4398.95	5	1	I	4683.83	8	40	II	4944.09	5	20	II
4402.30	4	20	II	4684.27	7	1	I	4944.96		7h	II
4409.35	6	10	II	4688.51	1	4h	II	4954.67		4h	II
4410.23	1?		I	4690.80	1	5	II	4956.17	2h		I
4411.10	5		I	4702.36	3		I	4956.49	2		I
4427.44	10h		I	4704.90	3		I	4966.91	100	4	I
4430.22	6	1	I	4712.82		30h	II	4972.30		3h?	II
4439.21	100	10	I	4718.66	20h		I	4973.90		2h?	II
4448.94	1?		I	4720.79	6	1	I	4974.16	10		I
4463.48		2	II	4726.08	60	250	II	4988.30	4	1	I
4467.14		2	II	4732.02	1	7	II	5009.53	20	50	II
4472.19		3	II	4732.96		3	II	5014.50		10	II

TABLE 1.—Arc and spark spectra of ytterbium (Z=70)—Continued

$\lambda$ air A	Intensity and character		Spectrum	$\lambda$ air A	Intensity and character		Spectrum	$\lambda$ air A	Intensity and character		Spectrum
	Arc	Spark			Arc	Spark			Arc	Spark	
5019.66	4	-----	I	5249.84	-----	8	II	5457.45	-----	2	II
5021.14	2	10	I	5254.21	-----	1	I	5458.22	-----	2h?	II
5027.66	10	-----	I	5255.68	-----	8	II	5471.21	2?	-----	II
5032.61	-----	2h?	II	5257.50	15	100	II	5474.05	10	-----	I
5033.33	-----	2h?	II	5258.19	5	-----	I	5478.52	10	50	II
5038.28	-----	3h?	II	5262.07	-----	1	II	5481.94	60	2	I
5047.12	-----	2h?	II	5263.61	-----	10	II	5486.59	-----	4	II
5049.87	2	30h	II	5275.58	3	-----	I	5493.10	1?	-----	I
5058.60	2	-----	I	5277.08	200	6	I	5498.86	-----	20hl	II
5058.76	-----	2h?	II	5279.56	15	100	II	5504.04	1?	-----	I
5062.95	-----	15	I	5280.53	-----	1	II	5505.50	40	2	I
5067.33	5	5	I	5286.14	-----	4	II	5508.30	-----	2h	II
5067.68	-----	3h?	II	5287.30	-----	5	I	5511.46	-----	2h	II
5067.80	10	-----	I	5299.86	2	-----	I	5518.39	1?	-----	I
5069.15	30	2	I	5300.95	6	60	II	5520.24	-----	4h	II
5074.33	200	5	I	5306.53	-----	1h	II	5524.55	10	-----	I
5074.68	-----	3?	II	5309.33	2	-----	II	5527.81	1?	-----	I
5076.75	50	1	II	5321.14	-----	6	II	5529.09	1	9	II
5081.00	3	-----	I	5328.51	-----	1h?	II	5529.95	-----	8h	II
5082.60	4	-----	I	5334.04	2h	-----	I	5539.09	200	5	I
5085.74	-----	4h	II	5335.15	150	400	II	5545.85	2?	-----	I
5087.64	1	10	II	5338.77	-----	3	II	5547.19	2	15	II
5088.97	-----	3h	II	5345.68	20	100	II	5548.97	-----	2h	II
5090.67	-----	2h	II	5345.82	10	50	II	5556.48	1500	50	I
5101.60	-----	4h	II	5347.21	40	200	II	5562.07	50	2?	I
5104.43	1	50	II	5351.33	50	3	I	5568.11	20	1	I
5105.06	-----	5	II	5352.96	100	250	II	5572.55	-----	6	II
5111.51	-----	6h	II	5357.12	-----	1	II	5573.14	1	-----	I
5113.34	3	-----	I	5358.65	15	100	II	5575.73	1	2	II
5116.23	-----	1h?	II	5359.98	-----	10	II	5578.23	5	-----	I
5117.75	-----	3h	II	5362.46	-----	2	II	5580.82	3	30	II
5121.61	1	15	II	5363.66	25	2	I	5585.43	2	-----	I
5126.80	4	-----	I	5364.12	-----	2	II	5586.35	20	-----	I
5128.54	-----	5h	II	5364.84	-----	1?	II	5587.29	1?	-----	I
5129.60	-----	3h	II	5365.44	-----	3	II	5588.46	30	100	II
5129.65	2	-----	I	5367.36	-----	2	II	5597.19	4	-----	I
5130.52	1	-----	I	5367.79	2	3	II	5598.48	15	-----	I
5132.41	-----	5h	I	5368.29	2	20	II	5600.03	2h?	-----	I
5136.00	6	50	I	5376.99	-----	10	II	5607.38	-----	2h	II
5139.53	3	-----	I	5380.26	2	-----	I	5608.92	-----	2h	II
5142.31	-----	5h	II	5380.57	3	-----	I	5620.17	2	-----	I
5147.02	3	50	II	5383.96	-----	2	II	5620.24	1	10	II
5152.34	-----	5h	II	5386.73	-----	2	II	5620.90	-----	1?	II
5173.13	1	15h	I	5388.01	5	-----	I	5627.89	-----	5	II
5178.73	3	-----	I	5389.87	8	30	II	5631.52	-----	3h	II
5180.36	2	10	II	5390.68	30	-----	I	5632.00	2	-----	I
5184.18	8	30	II	5390.85	30	-----	I	5637.80	2	-----	I
5187.38	2	-----	I	5393.40	-----	6	II	5637.86	-----	5	II
5193.87	4h	-----	I	5393.85	6hl	-----	I	5639.35	1?	-----	I
5194.76	3h?	-----	I	5395.78	-----	7	II	5652.00	50	80	II
5196.09	20	3	I	5399.71	2	-----	I	5653.29	-----	5	II
5200.57	-----	10	II	5399.76	-----	15	II	5654.33	-----	2	II
5211.59	40	2	I	5403.14	20	1	I	5663.19	1	-----	I
5215.45	-----	5h	II	5409.48	-----	2	II	5669.98	1?	-----	I
5217.98	-----	7h	II	5414.29	1	20	II	5683.60	2	6	II
5226.19	-----	4	II	5424.68	-----	10h	II	5686.53	5	-----	I
5227.09	-----	3?	I	5426.91	2	60	II	5686.57	?	10h	II
5227.25	10	-----	I	5429.86	-----	1h?	II	5689.71	-----	1h	II
5228.18	5	-----	I	5431.20	-----	2	II	5689.92	10	-----	I
5229.99	-----	2?	II	5432.73	5	100	II	5693.71	1	10	II
5236.68	-----	7h	II	5440.58	-----	5	II	5705.00	-----	2	II
5240.51	10	40	I	5445.52	-----	1	II	5706.79	-----	1?	II
5244.11	50	5	I	5449.30	20	100	II	5713.75	1	10	II
5244.65	-----	10	II	5454.01	5	-----	II	5717.30	-----	15h	II
5246.41	-----	1?	II	5455.08	1	6	II	5720.01	300	8	I

TABLE 1.—Arc and spark spectra of ytterbium (Z=70)—Continued

$\lambda$ air A	Intensity and character		Spec- trum	$\lambda$ air A	Intensity and character		Spec- trum	$\lambda$ air A	Intensity and character		Spec- trum
	Arc	Spark			Arc	Spark			Arc	Spark	
5723.72	3		I	6001.05		1?	II	6312.99		1h?	II
5728.86	5		I	6002.52		2?	II	6324.42		1?	II
5730.02	4	60	II	6003.65	5		I	6324.73		2	I
5735.80	2	15	II	6004.53	2?		I	6324.78	3		I
5736.95	3?		I	6007.42	4	80h	II	6333.36		1?	II
5745.80	6		I	6014.95	3		I	6335.72	8		I
5749.92	8	10	II	6020.56	2	5	II	6345.02	15h		I
5754.79		1h?	II	6021.97		15h	II	6345.74	2	7	II
5755.90	15		II	6024.08	2	5	II	6355.40	2	50h	II
5764.65		1h?	II	6031.80	10		I	6356.68		2h?	II
5767.23	2	10h	II	6035.72	2		I	6372.71	6		I
5771.67	30	50	II	6040.80	3	10	II	6377.04		6h	II
5773.86	2		I	6042.26		2?	II	6382.93	1	2	II
5774.35	2		I	6048.43	15		II	6387.72	3		I
5775.68		1?	II	6052.88	8	30	II	6393.79	2?		I
5778.00		3	II	6054.56	10		I	6400.40	200h	4h	I
5786.60		6	II	6056.48	1	20h	II	6402.64		1h?	II
5789.94	5		I	6059.25	10		I	6404.62	2h		I
5799.59		2h?	II	6075.25		3	I	6417.97	120	3	I
5803.45	15		I	6082.40		10h	II	6421.02	1?		I
5808.69		2	II	6083.29		2	II	6421.54	3		I
5810.68	2?		I	6083.85		1?	II	6432.73	30	40	II
5811.95		1?	II	6088.72		3	II	6438.01		1h?	II
5819.43	7	100	II	6098.51		1	II	6440.81		5	II
5824.35		2?	II	6106.22		4	II	6456.92		4	II
5831.82	1		I	6110.19		4	II	6463.15	10	100	II
5834.00	60h	1	I	6111.28	7		I	6474.74	5	50	II
5834.58	1?		I	6118.40	10h?		I	6488.50		1	I
5835.54		1?	II	6120.38		20h	II	6489.10	800	40	II
5836.22		2?	II	6122.99	4		I	6489.27	?	30	II?
5837.15	50	150	II	6123.03		4	II	6492.74	3	50	II
5838.67		2	II	6126.37		2?	II	6503.02	2	4	II
5842.45	1?		I	6128.21		6	II	6509.11	2?		I
5844.46	2?		I	6131.61		2h?	II	6523.24		4?	II
5847.24		2h	II	6134.31		5	II	6541.37	1?		I
5854.52	30	1	I	6142.85		1?	II	6550.19	10	?	I
5865.65		2	I	6146.94		15h	II	6553.34	2	?	I
5868.40		6	II	6150.64	30h		I	6555.15	3	?	I
5874.70	1	30h	II	6152.58	60	80	II	6568.35	2	?	I
5875.96	2?		I	6171.65		2	II	6585.42	7	20	II
5882.81	2	10	II	6175.59		1	II	6592.75		3?	II
5896.61	5		I	6189.07		1?	II	6605.93		4h	II
5897.22	7	100h	II	6190.81	3	15	II	6607.07	20		I
5898.80	3	50h	II	6194.85	3		I	6617.06	10	6	I
5903.37		2	II	6208.11	2	20h	II	6643.54	50		I
5908.36	20	30	II	6208.68		1?	II	6644.07	2	5h	II
5920.39	1	10	II	6215.57		6	II	6661.90		4h?	II
5927.70	2		I	6223.65		15	II	6667.85	1000	20	I
5935.06	3	40	II	6233.38		5	II	6678.17	20		I
5936.47	2		I	6234.86		5	II	6679.32		4?	II
5943.55	1	2	II	6235.27	2		I	6692.42	2?		I
5946.02	4	100	II	6236.15		2?	II	6699.38	10	8	II?
5947.28	1	5	II	6236.56	2		I	6715.79	5		I
5950.65	3		I	6246.97	40	60	II	6727.62	30	60	II
5950.98		8	II	6247.99	3		I	6745.22	2	7	II
5955.35	2		I	6260.80	4	50h	II	6749.40	7		I
5958.70	10		I	6265.46		1?	II	6755.42		4?	II
5959.31	3		I	6270.33		1?	II	6768.70	80	2	I
5961.85		2h?	II	6271.16	1?	3	II	6777.22	2		I
5972.71	1?		I	6274.79	100	150	II	6782.17	4h		I
5985.35	1	5	II	6277.13		5	II	6785.16	1	5	II
5986.80		2h?	II	6286.26	8		I	6790.82	1?		I
5987.91	2	20h	II	6297.38		15h	II	6799.61	1000	50	I
5989.32	15		I	6303.28	3	4	II	6802.47	2	20	II
5991.51	50	150	I	6308.16	20	30	II	6813.66	1?		I

TABLE 1.—Arc and spark spectra of ytterbium (Z=70)—Continued

$\lambda$ air A	Intensity and character		Spec-trum	$\lambda$ air A	Intensity and character		Spec-trum	$\lambda$ air A	Intensity and character		Spec-trum
	Arc	Spark			Arc	Spark			Arc	Spark	
6820.04	3h	-----	I	7175.14	10	-----	-----	8053.41	10	-----	-----
6826.06	-----	2?	II	7187.06	5	-----	-----	8607.51	8	-----	-----
6828.90	1	6	II	7222.72	3	-----	-----	8922.61	20	-----	I
6833.57	2h?	-----	I	-----	-----	-----	-----	9094.44	3	-----	-----
6864.26	-----	10h?	II	7244.47	20	-----	-----	9104.06	3	-----	I
6871.54	10	-----	I	7251.5	2?	-----	-----	9304.44	15	-----	-----
6877.94	-----	8h?	II	7305.25	15	-----	-----	9349.27	20	-----	-----
6878.12	3?	-----	I	7313.10	20	-----	-----	9392.8	5	-----	-----
6881.43	-----	4h?	II	7342.3	2?	-----	-----	9524.43	4	-----	-----
6889.60	-----	4?	II	7350.09	40	-----	-----	9677.8	2?	-----	-----
6913.55	2?	-----	I	7361.8	1?	-----	-----	9688.6	3?	-----	-----
6934.04	15	10	I, II?	7377.55	4	-----	-----	9760.37	100	-----	-----
6951.51	2	-----	I	7405.92	3	-----	-----	9799.88	10	-----	-----
6963.11	2	5	II	7422.1	4?	-----	-----	9870.07	6	-----	-----
6977.68	2	-----	I	7448.33	30	-----	-----	9976.40	4	-----	-----
6982.00	3	2	I, II?	7527.56	80	-----	I	10267.36	7	-----	-----
6999.87	15	3	I	7699.49	1500	-----	I	10321.64	5	-----	-----
7020.16	5	-----	-----	7722.58	2	-----	-----	10770.1	2Lu?	-----	-----
7027.78	2?	-----	-----	7734.5	1?	-----	-----	-----	-----	-----	-----
7032.04	3	-----	-----	7758.03	10	-----	-----	-----	-----	-----	-----
7032.91	2	-----	-----	7781.5	1?	-----	-----	-----	-----	-----	-----
7043.79	8	-----	-----	7877.30	15	-----	-----	-----	-----	-----	-----
-----	-----	-----	-----	7895.12	20	-----	-----	-----	-----	-----	-----
-----	-----	-----	-----	7922.40	7	-----	-----	-----	-----	-----	-----

The data of table 1 represent at least three different spectra, each characteristic of Yb in some form or another. About 400 lines belong to neutral Yb atoms (Yb I spectrum), approximately 1,250 characterize singly ionized atoms (Yb II spectrum), and possibly a dozen belong to doubly ionized atoms (Yb III spectrum). We found no evidence of band heads which could be assigned to a molecular spectrum of ytterbium compounds.

The successive atomic spectra are differentiated mainly on the basis of intensity comparisons of arc and spark spectra, but since the latter were not recorded beyond 7,000 Å the lines of longer wave length from ionized atoms remain unobserved, or unrecognized if recorded in arc spectra. The Yb I spectrum is surprisingly simple, but may not be fully developed in the arc. It may possess more lines in the infra-red but the available Yb material was too scarce and precious to permit long exposures required to photograph beyond 10,500 Å. The Yb II spectrum is probably even more complex than is apparent because in addition to being undetermined beyond 7,000 Å some lines are probably overlooked on account of being masked by silver lines or by the air spectrum.

Our assignment of lines to successive spectra agrees almost perfectly with King's [13] separation of ionization stages based upon a study of the electric-furnace spectra of Yb. This grouping of lines is further confirmed by the spectral-term analyses, insofar as they have been carried out.

Since no details concerning regularities in the spectra of ytterbium have heretofore been published, we present a preliminary report, as follows: Soon after making our first measurements and identifications of lines from neutral Yb atoms, in 1930, we found from repeated differences among wave numbers three energy levels with separations of 703.5 and 1718.4  $\text{cm}^{-1}$ . These were interpreted as  $(f^{14}sp)^3P_{0,1,2}$  by

Prof. H. N. Russell, who promptly identified two singlet levels ( $f^{14}s^2$ ) $^1P_1$  and ( $f^{14}s^2$ ) $^1S_0$ , which account for the two lines of outstanding intensity and easy excitation in the Yb I spectrum. These terms and their combinations are displayed in table 2 where the term symbols and relative values appear on the margin and the wave lengths (intensities) and wave numbers of the observed combinations at intersections of lines and columns. Only the terms which can be interpreted with considerable certainty are given now, the remainder being reserved until the analysis is completed and confirmed by observations of Zeeman effects, which, up to the present time, have not been produced for Yb spectra. It appears certain that the normal state of neutral Yb atoms is described by ( $4f^{14}6s^2$ ) $^1S_0$  and that the most intense line of the Yb I spectrum is ( $4f^{14}6s^2$ ) $^1S_0$ —( $4f^{14}6s6p$ ) $^1P_1$  with wave length 3,987.99 Å. This spectrum closely resembles that of an alkaline earth. Indeed, the fact that Yb may be bivalent accounts for its separation by electrolytic reduction from other trivalent rare earths [14].

Proceeding in a similar manner with Yb II lines the authors found several score of levels combining to produce some hundreds of lines, but it was impossible to group the levels into terms and identify them. Two lines of extraordinary intensity featuring the Yb II spectrum were not included among our level combinations and Professor Russell suggested that they represent the transition ( $f^{14}s$ ) $^2S_0$ —( $f^{14}p$ ) $^2P_{1,2}$ . When the  $f$  shell is completely filled ( $f^{14}$ ) it contributes nothing to the spectrum and a single-valence electron will produce an alkali-like spectrum with ( $s$ ) $^2S_0$  describing the normal state. However, the relatively great complexity of the Yb II spectrum indicates that most of the excited states must arise from electron configurations of types  $f^{13}s^2$ ,  $f^{13}sd$ ,  $f^{13}d^2$ ,  $f^{13}sp$ ,  $f^{13}sd$ , which produce large families of spectral terms. This is probably the explanation of the numerous levels first discovered, but further attempts to interpret them and connect them with terms arising from configurations with  $f^{14}$  electrons will be postponed until Zeeman effects are available.

TABLE 2.—Terms and combinations in the Yb I spectrum

Term symbol ----- Value -----		( $4f^{14}6s6p$ ) $^3P_0^o$ 17288. 5	( $4f^{14}6s6p$ ) $^3P_1^o$ 17992. 0	( $4f^{14}6s6p$ ) $^3P_2^o$ 19710. 4	( $4f^{14}6s6p$ ) $^1P_1^o$ 25068. 2
Term symbol	Value				
( $4f^{14}6s^2$ ) $^1S_0$ -----	0. 0		5556. 48(1500) 17992. 0		3987. 98(2000) 25068. 2
( $4f^{14}6s7s$ ) $^3S_1$ -----	32694. 7	6489. 10(800) 15406. 2	6799. 61(1000) 14702. 7	7699. 49(1500) 12984. 3	
( $4f^{14}6s5d$ ) $^3D_1$ -----	39808. 7	4439. 21(100) 22520. 2	4582. 36(50) 21816. 7	4974. 16(10) 20098. 3	6782. 17(4) 4740. 5
( $4f^{14}6s5d$ ) $^3D_2$ -----	39838. 0		4576. 21(200) 21846. 0	4966. 91(100) 20127. 6	6768. 70(80) 4769. 8
( $4f^{14}6s5d$ ) $^3D_3$ -----	39965. 1			4935. 51(500) 20255. 7	
( $4f^{14}6s5d$ ) $^1D_2$ -----	40061. 5		4529. 90(20) 22069. 4	4912. 38(20) 20351. 1	6667. 85(1000) 4993. 2
( $4f^{14}6s8s$ ) $^3S_1$ -----	41614. 9	4109. 60(5) 24326. 4	4231. 99(10) 23022. 9	4564. 00(50) 21904. 5	
( $4f^{14}6s6d$ ) $^1D_2$ -----	44760. 3		3734. 70(6) 26768. 3	3990. 89(40) 25050. 0	5076. 75(50) 19692. 2

The resonance lines of Yb II are interpreted with considerable certainty, as shown in table 3. The most intense line emitted by singly ionized Yb atoms is expected to be  $(f^{14}6s)^2S_{1/2} - (f^{14}6p)^2P_{1/2}$  with wave length 3289.36 Å, but the estimated intensities appear to be somewhat anomalous. Apparently 3694.19 Å is the strongest Yb II line.

No attempt has been made to analyze the Yb III spectrum, but it may be expected that the ground state of doubly ionized Yb atoms is described by  $(4f^{14})^1S_0$ . The full shell of *f*-type electrons is likely to have high stability on account of which the fundamental lines emitted by  $Yb^{++}$  ions will probably lie in the extreme ultraviolet.

TABLE 3.—Terms and combinations in the Yb II spectrum

Term symbol..... Value.....		$(4f^{14}6p)^2P^{\circ}_0 \frac{1}{2}$ 27061.9	$(4f^{14}6p)^2P^{\circ}_1 \frac{1}{2}$ 30392.3
Term symbol	Value		
$(4f^{14}6s)^2S_0 \frac{1}{2}$	0.0	3694.19(1000) 27061.9	3289.36(800) 30392.3
$(4f^{14}7s)^2S_0 \frac{1}{2}$	54304.3	3669.71(10) 27242.4	4180.82(100) 23912.0

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